This article uses a business process renovation method to address an environmental protection problem. The presented case, studied in factories which remake crude vegetable oil (soapsuds remake), focuses on reengineering one of the business processes which is classified as an obligatory process. This process has to be performed to comply with the requirements of the existing ecology legislation. Therefore, it is reasonable to take a radical approach - business process reengineering. Soapsuds with oil remnants are a secondary product in the vegetable oil refining process. According to the legislation, secondary products as waste are not acceptable for the environment. Soapsuds remake as it is currently carried out produces technical fatty acids and, as a side product, calcium sulphate. Calcium sulphate is listed as special waste; therefore it must be deposited in a special waste landfill site. In the course of searching for a solution to this ecological problem, a new idea came up, namely that soapsuds are taken to the biogas plant. At the biogas plant, they can be decomposed into biogas, which is then used to generate electric energy, for heating or to supplement municipal gas. Thus, the reengineering of the process is upgraded into process innovation and environmental innovation. Using data obtained from the available literature, we managed to prove that process reengineering, which is simultaneously process innovation and ecological innovation, can be applied in practice. Side products resulting from the anaerobic digestion of soapsuds in the biogas plant comply better with the requirements of the Slovenian ecological legislation than those occurring in the process used up to now.

**Key words:** business process reengineering, process innovation, soapsuds remake, biogas plant, ecology

### 1 Introduction

The constant new demands of the market and fast technological development require that organisations increase value for the customer, cut costs and stay competitive. In addition, they have to comply with legislative requirements, in particular with the ecology legislation, which is becoming more and more stringent.

We are living in the age of globalization; therefore companies need to adjust to the new conditions. The efficiency of an organization is based on timely and efficient adjustment to market conditions, new product conditions, economic factors and the social environment. In the past, the main form of an organization was its structure, something which has not changed significantly over time. However, today it is a key process which is constantly changing. Here we have information about what goes on within a company (Vila, 2009).

Too wide a gap between the structure and processes results in an inefficient business system. The productivity and commercial success of an organization depend on its harmonization of key production factors: human resources, technology and organization of work.

Therefore, today major emphasis is placed on business process reengineering, which can gradually, continually or radically transform business processes.

Companies also endeavour to achieve a competitive advantage by investing into equipment, technology and development, with varying degrees of success. The reasons for technological changes in companies are usually a combination of the following factors (Frankel, 1990):

- market pressure to improve products and services;
- competitive pressure and competition on the market;
- the company’s endeavours to increase production;
- scientific developments;
- moral factors, e.g. national pride, and
- coincidences (innovations and inventions, etc.)

The fact that society, social groups and individuals have to continually prove themselves by coming up with innovations and new products, which is not only very interesting, but also commercially viable on the commodities market. Therefore, in order to find a good idea, we need not only theoretical knowledge, but also experience and practice.

Ecology has played an increasingly important role in recent times. With regard to protect environment due to global environmental changes and, consequently, the ever stricter ecology legislation, we can conclude that today’s market
Figure 1: Organizational structure - model (Verbič, 2008).
The processing of crude vegetable oil is continuous. One of the phases is neutralization of the oil, wherein by means of phosphoric (V) acid, $\text{H}_3\text{PO}_4$, non-hydratable phosphatides are removed (degumming of crude oil). With sodium hydroxide, NaOH, saturated and unsaturated free fatty acids are removed (neutralization of crude oil). These processes produce soap-suds (aqueous soap solution) as a secondary product. They are separated from the oil in a separation process based on various densities (oils are lighter than aqueous soap solutions). This process is necessary because free fatty acids decompose within oil and contribute to rancidity which causes rapid deterioration of the oil.

Soapsuds remake (soap splitting), as presented in this article, is considered an obligatory process. It must be performed so as to comply with the ecology legislation.

Soap splitting is either a continuous process or a process which takes place in batches (batch soap splitting). In Slovenia, batch soap splitting is more common, therefore it is this process that will be described here (Verbič, 2008, Verbič and Kern, 2008).

An aqueous soap solution is gathered in a special batch, in a room which is physically separated from other parts of the refinery (one or more batches). After the batch has been filled, the worker in charge starts with the process. The reaction takes place at a temperature of 80 – 90°C. During intensive mixing, which takes place for 3 hours, concentrated sulphuric acid $\text{H}_2\text{SO}_4$ is added to the solution. This reaction reproduces technical fatty acids which separate from the solution (they are lighter than the aqueous solution). The techni-
CURRENT PROCESS - MODEL (AS-IS)

![Diagram of the current process model]

Figure 2: Soapsuds remake - model (Verbič, 2008).
Cal fatty acids are then filtered and washed with water (Swern, 1964). The laboratory takes a sample to test for any remnants of sulphuric acid and water in the technical fatty acids. The results of the laboratory analyses are recorded in the electronic analysis logbook. The refinery line manager and the worker in the room where the batch soap splitting takes place, have access to the electronic analysis logbook. If the results of the analysis comply with the production specifications, the worker pumps the technical fatty acids into a reservoir; if they do not, the process of washing, sedimenting and sampling is repeated. The refinery line manager makes an entry in the refinery work logbook (and in the electronic logbook) recording the amount of sulphuric acid used, which is kept in reservoirs.
The refinery line manager also monitors the stock of sulphuric acid (he/she records this data electronically and submits a purchase order via intranet to the Purchasing Department). The purchasing clerk completes the purchase. The refinery line manager takes delivery of the package of sulphuric acid, records the quantity in the electronic logbook, signs the delivery note and sends it to the Purchasing Department, wherein the procedure (after receipt of the invoice) is completed.

The refinery line manager also monitors the quantity of fatty acids in the reservoirs. When the amount is sufficient, the refinery line manager orders the sale of the technical fatty acids (for the cosmetic industry) to the Sales department clerk. The Sales department clerk completes the sales procedure. After a warehouse worker has pumped the technical fatty acids into a cistern vehicle, the head of the warehouse fills out the dispatch note (prepared by the Sales department clerk) with information about the quantity of technical fatty acids to be shipped and sends the dispatch note to the Sales department clerk. The laboratory clerk receives a written document with an analysis signed by the head of the laboratory and, together with the dispatch note and the invoice, submits them as accompanying documents (Verbič, 2008, Verbič and Kern, 2008).

The second part involves wastewater. The resulting wastewater - the aqueous solution of sodium sulphate (\(\text{Na}_2\text{SO}_4\)) and the remnants of sulphuric acid, soap, phosphates and neutral oil – the fatty phase, is piped into sedimentation chambers (where the remnants of the fatty phase are separated from the wastewater). From there, the fatty phase continues to the batch soap splitting, while the wastewater continues to the wastewater treatment plant (Swern, 1964). Usually, companies have their own waste water treatment plants.

The aforementioned wastewater is collected in the equalisation pool of the waste water treatment plant. Sulphates are removed from the wastewater using lime. Solid calcium sulphate, \(\text{CaSO}_4\), is precipitated (with the remnants of the substances listed above). It is removed from the wastewater by the addition of a flocculant and transported to a special waste landfill site, as this waste still a special waste (Figure 6).

The procedure used to purchase lime and flocculant is identical to the purchase of sulphuric acid, except that it is done by the line manager of the waste water treatment plants. The process of dispatching the calcium sulphate is completed by the Sales department clerk, who prepares the dispatch note and receives the invoice for transport and deposition (Verbič, 2008, Verbič and Kern, 2008).

Technical fatty acids are a product which brings added value, but also incurs costs during production: the cost of the sulphuric acid and lime, work costs, costs of transporting and depositing the calcium sulphate in the special waste landfill site, energy costs, investment into equipment, tax for pollution of the environment. The process of soapsuds remake is presented in the models given below (Figures 2 and 3).

### 2.4 Description of schedule for individual functions

The schedule of the process is given in the following table.

<table>
<thead>
<tr>
<th>Function</th>
<th>AWAITING EXECUTION</th>
<th>PREPARATION FOR EXECUTION</th>
<th>EXECUTION</th>
<th>FINALISATION</th>
</tr>
</thead>
<tbody>
<tr>
<td>SOAP SPLITTING</td>
<td>cca 8 hours until the batch is full</td>
<td>30 minutes</td>
<td>4 hours*</td>
<td>5 minutes</td>
</tr>
<tr>
<td>LABORATORY – ANALYSIS OF TFA</td>
<td>10 minutes</td>
<td>7 minutes</td>
<td>10 minutes</td>
<td>5 minutes</td>
</tr>
<tr>
<td>REPUMPING TFA INTO THE RESERVOIR</td>
<td>5 minutes</td>
<td>2 minutes</td>
<td>15 minutes</td>
<td>5 minutes</td>
</tr>
<tr>
<td>REPUMPING WATER INTO THE CLEANING PLANT</td>
<td>5 minutes</td>
<td>5 minutes</td>
<td>15 minutes</td>
<td>5 minutes</td>
</tr>
<tr>
<td>PURCHASE OF SULPHURIC ACID, LIME AND FLOCULANT</td>
<td>5 minutes</td>
<td>5 minutes</td>
<td>10 minutes</td>
<td>5 minutes</td>
</tr>
<tr>
<td>DELIVERY OF CHEMICALS</td>
<td>5 minutes</td>
<td>3 minutes</td>
<td>30 minutes</td>
<td>10 minutes</td>
</tr>
<tr>
<td>TFA SALES PROCEDURE</td>
<td>for the reservoir to be full</td>
<td>30 minutes</td>
<td>30 minutes</td>
<td>10 minutes</td>
</tr>
<tr>
<td>ANALYSIS OF TFA FROM THE COLLECTION RESERVOIR</td>
<td>10 minutes</td>
<td>7 minutes</td>
<td>10 minutes</td>
<td>10 minutes</td>
</tr>
<tr>
<td>PUMPING TFA INTO A CISTERN VEHICLE FOR TRANSPORT</td>
<td>10 minutes</td>
<td>7 minutes</td>
<td>25 minutes</td>
<td>7 minutes</td>
</tr>
<tr>
<td>FINAL PROCEDURE IN SALES DEPARTMENT</td>
<td>5 minutes</td>
<td>5 minutes</td>
<td>10 minutes</td>
<td>10 minutes</td>
</tr>
<tr>
<td>WASTEWATER TREATMENT</td>
<td>5 minutes</td>
<td>5 hours</td>
<td>15 minutes</td>
<td></td>
</tr>
<tr>
<td>TOTAL</td>
<td>55 minutes</td>
<td>1 hour 46 minutes</td>
<td>11 hours 25 minutes</td>
<td>1 hours 57 minutes</td>
</tr>
</tbody>
</table>

**NOTES:** * 0.5 hour for heating, 3 hours for batch soap splitting and 0.5 hour for washing
The times for the individual process stages are long, but they cannot be significantly shortened with the same chemical procedures (Verbič and Kern, 2008).

3 The analysis

3.1 Analysis of the process structure

The process itself is rather complicated, as it includes 19 workers (from the purchasing department, sales department, refinery, warehouse, cleaning plant and laboratory). As regards the number of people involved in the process, we took account of the fact that the batch soap splitting takes place during three shifts, whereas wastewater treatment during two shifts. The procedure is complicated and is performed in three organizational units and at three levels within the organizational units (Figure 1). The process itself should be simplified, i.e. it should involve fewer workers who would perform a greater number of consecutive tasks. There is no technical support for monitoring the quantity of chemicals and technical fatty acids in reservoirs. If this was available, the waiting and preparation time would be shortened however we cannot significantly shorten the technological time with the same kind of chemical procedures (Verbič and Kern, 2008).

3.2 Time analysis of the process

In terms of time, it takes around 16 hours to perform this process. It is impossible to significantly shorten the technological time for performing the process (which is the most time consuming) with the same kind of chemical procedures.

3.3 Analysis of influential factors

With regard to the ecological legislation currently applicable both in Slovenia and in the European Union, and the ecological policies of Slovenia and the European Union, both of which support sustainable development, it seems clear that environmental protection has become more of an important issue. We can expect that tax for pollution of the environment will increase, making it necessary to find solutions which will be financially more beneficial for the business system and will, at the same time, comply with the ever more stringent ecological legislation.

3.4 Comparative analysis

Business systems whose main activity is vegetable oil processing (through alkaline refining) are dealing with this problem in a similar manner. The difference lies in the fact that some of them opt for batch soap splitting, while others apply a continuous soap splitting, while wastewater is treated either in their own wastewater treatment plants or they pay for it to be treated in communal wastewater treatment plants.

4 Proposal for reengineering and innovation of the process

4.1 The process after reengineering and innovation

Since a company has to perform this process, its aim is to do as cheaply as possible or perform it in a way which proves the most beneficial in terms of added value to the business system.

Business process reengineering offers this possibility. While searching for a solution, it was proposed that soapsuds should no longer be remade, but taken to a biogas plant instead (Figure 7). At the biogas plant they can be decomposed into biogas, which is then used to generate electric energy, for heating or to supplement municipal gas. Thus, process reengineering is both process innovation and ecological innovation.

The scope of the process would be minimized, which also presents a change in the process on the first, second and third levels. It would no longer be necessary to perform laboratory analyses and purchase sulphuric acid. Consequently, the use of lime (currently, approximately 1 t is used to process 100 t of crude vegetable oil) and flocculant would decrease. This means much less paperwork is required for this kind of task (sample analyses, purchase orders, delivery notes, etc.), fewer workers involved in the process (reduction from 19 to 3 workers) and less time needed to perform the tasks (cut from 16 hours 3 minutes to 3 hours 25 minutes).

The company would collect soapsuds using the existing reservoirs for technical fatty acids and for sulphuric acid. When a suitable amount of soapsuds is collected (25 t is the amount that can be transported in a cistern vehicle), the soapsuds would be transported from the company to the biogas plant. The warehouse worker would, following the order from the line manager of the warehouse, pump the soapsuds into the cistern vehicle, while the line manager of the refinery would handle the dispatch procedure. He prepares the dispatch note and signs it together with the head of the warehouse; the document is then submitted as an accompanying document. One copy is sent to the purchase department for information. The process is completed when the company receives the invoice for the procedure. This change would require additional training for the line manager of the refinery (Verbič and Kern, 2008). The model of the changed process is shown in Figure 4.

4.2 Organizational and human resources structure after the reengineering and innovation of the process

Innovation and process reengineering result in changes in the organizational structure, i.e. they decrease the number of organizational units involved in the process (Figure 5). The human resources system is also altered. The systematization of work positions would change, in particular systematization of the position handling the batch soap splitting - the worker would be able to perform other assignments (e.g. the work
**PROCESS REENGINEERING - MODEL (TO BE)**

![Diagram of process reengineering](image)

*Figure 4: Soapsuds remake after process reengineering - model (Verbić, 2008).*

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**LEGEND:**

- Involved in the studied process

*Figure 5: Organizational units involved in the reengineered process - model (Verbić, 2008).*
of a stoker) in half the time. Systematization of work positions for the refinery line manager, purchasing department clerk and workers in the wastewater treatment plant would also change. The proposed changes require modification of the IT system. Therefore, it would be sensible to implement computer-supported measurement of the amount of soapsuds in the reservoirs. The level of soapsuds would be measured by means of an electronic level measuring device. The signal would be connected to the computer. This would enable the refinery line manager to check the amount of soapsuds at any time. This would additionally simplify the process (Verbič and Kern, 2008).

### 4.3 Schedule for individual functions after process reengineering and innovation

The schedule of individual functions after the change is as follows (Verbič and Kern, 2008):

<table>
<thead>
<tr>
<th>Function</th>
<th>Waiting Execution</th>
<th>Preparation For Execution</th>
<th>Execution</th>
<th>Finalisation</th>
</tr>
</thead>
<tbody>
<tr>
<td>Repumping soapsuds into the reservoir</td>
<td>5 minutes</td>
<td>10 minutes</td>
<td>15 minutes</td>
<td>10 minutes</td>
</tr>
<tr>
<td>Soapsuds sales procedure</td>
<td>for the reservoir to be full</td>
<td>30 minutes</td>
<td>30 minutes</td>
<td>10 minutes</td>
</tr>
<tr>
<td>Pumpeing TFA into a cistern vehicle for transport</td>
<td>10 minutes</td>
<td>15 minutes</td>
<td>25 minutes</td>
<td>15 minutes</td>
</tr>
<tr>
<td>Final sales procedure</td>
<td>5 minutes</td>
<td>5 minutes</td>
<td>10 minutes</td>
<td>10 minutes</td>
</tr>
<tr>
<td><strong>Total</strong></td>
<td><strong>20 minutes</strong></td>
<td><strong>1 hour</strong></td>
<td><strong>1 hours 20 minutes</strong></td>
<td><strong>45 minutes</strong></td>
</tr>
</tbody>
</table>

It is clear that the reengineered and innovated process significantly cuts waiting, preparation and finalisation times, but in particular the time spent executing the process itself. Total time spent decreases from the estimated 16 hours 3 minutes to 3 hours 25 minutes.

### 5 Estimated results of the anaerobic digestion of soapsuds in the biogas plant

In order to implement this reengineered and innovated process (which is also innovative in terms of ecological protection) in practice, it is important to know what effect the soapsuds will have on the operation of the biogas plant, as well as how the products resulting from the anaerobic digestion will affect the environment.

Drawing on information from the available literature, we will prove that soapsuds can actually be decomposed in a biogas plant and that the products (parameters) obtained from decomposition are more acceptable for the environment than products generated by the existing process.

We will describe the effect of the individual substances (parameters) which compose soapsuds on the operation of the biogas plant and, after decomposition, on the environment, both in terms of the pollution of water and soil. The effect will be shown by taking into consideration the concentration of these substances in soapsuds, but without considering the dilution which will most likely occur (depending on the composition of other substrates which are decomposed in the biogas plant). In a larger biogas plant many substrates (up to 250 t/day) are decomposed at the same time, not just soapsuds (Verbič, 2008).

#### 5.1 Composition of soapsuds

We determined the composition of soapsuds on the basis of data obtained from the available literature and using calculations based on the data. We took into consideration the addition of sodium hydroxide NaOH and phosphorus (V) acid $\text{H}_3\text{PO}_4$ in the crude oil degumming and neutralization phase. Soapsuds as a secondary product of crude vegetable oil processing thus contain (Verbič, 2008):

- in addition to soap, obtained from free fatty acids with added NaOH and Na ions, it also contains phosphatides, phosphate (added phosphorus (V) acid) to transform non-hydratable into hydratable phosphatides), several other non-defined gums and remnants of neutral oil.
- Data on the composition of soapsuds, expressed in % by weight (Rac, 1964):
  - unsaponifiable matters (mostly phosphatides) 0.34%
  - neutral oil up to 6%
  - soap approximately 13.66%
  - water around 80% (depending how much water is used for washing oil when removing soap).

#### 5.2 Total content of substances in soapsuds

When processing 100 t of crude vegetable oil, approximately 7 t of soapsuds are produced (depending how much water is
Figure 6: Scheme of batch soap splitting and wastewater treatment (Verbič, 2008).

Figure 7: Scheme of anaerobic digestion of soapsuds in the biogas plant (Verbič, 2008).
used for washing oil) which, according to the calculations, contains (Gunstone et al, 1986, Verbič, 2000, Verbič, 2008):

- Soap: 1,000 kg
- Sodium (in soap): 81.9 kg
- Neutral oil: 420 kg
- Phosphatides: 105 kg
- Phosphorus in phosphatides: 0.26 kg
- Phosphorus (added H$_3$PO$_4$): 31.6 kg
- Total phosphorus (added H$_3$PO$_4$ and phosphatides): 31.9 kg
- Water: 5443.4 kg

### 5.3 The biogas plant

In the biogas plant, decomposition of organic substances takes place under anaerobic conditions and as a result of the action of anaerobic microorganisms, which in an oxygen-free environment – i.e. anaerobic conditions – decompose organic materials into inorganic products (CO$_2$, H$_2$, H$_2$S, and ammonium - NH$_3$) and methane CH$_4$.

In the anaerobic process under optimum conditions the proportion of organic substances present can be reduced by 75%. The anaerobic digestion of organic substances can be shown with the following equation:

\[
\text{Organic matter} \xrightarrow{\text{Anaerobic fermentation bacteria}} \text{CH}_4 + \text{CO}_2 + \text{H}_2 + \text{H}_2\text{S} + \text{new bacteria} + Q
\]

From the equation we can see that, in the absence of oxygen, anaerobic fermentation bacteria decompose organic matter into the gaseous products CH$_4$ and CO$_2$, H$_2$ and that simultaneously a certain amount of new bacterial mass is produced with an equivalent amount of organic matter. Part of the organic matters is transformed into thermal energy (Baras et all, 1979). On the one hand, we obtain biogas, while on the other we get a stable sludge which is easier to store than sludge from aerobic wastewater treatment plants. The resulting wastewater is usually sent in the aerobic wastewater treatment plants.

The anaerobic digestion is always only a part of the entire cleaning procedure (75% reduction of the organic matters), but with a well guided aerobic digestion we can reduce the organic matters by 95%. It is important that with anaerobic digestion high concentrations of organic materials in wastewater do not restrict operation, but for aerobic digestion high concentrations of organic materials in wastewater (BPK5 higher than 2000) are a limitation due to the speed of oxygen transfer from the gaseous to the liquid phase (Baras et al., 1979). The latest practical experience shows that it is possible to reduce the organic matters by up to 90% with a well guided anaerobic digestion.

The use of the anaerobic digestion is therefore wholly applicable when cleaning industrial sewage with organic matters when one or two of the following conditions are met (Baras et all, 1979):

- high concentration of organic matter in water,
- high water temperature.

The organic compounds decompose (in wastewater) until CO$_2$ and methane are produced (desired products in the decomposition), as well as certain metabolites following this scheme (Figure 8) (Bischofsberger et al., 2005):

1. In hydrolysis, undissolved compounds are cracked into monomers by exoemnzymes (hydrolaze).

![Figure 8: The anaerobic degradation of complex organic compounds (Bischofsberger et al., 2005).](image-url)
2. The monomers (monosacharides, amino acids, fatty acids) formed in the hydrolytic phase are taken up by different facultative and obligate anaerobic bacteria and are degraded in the second, the acidogenic phase, to short chain organic acids. Depending on the environment and the substrate, different products are produced: in the water phase these are mostly lower organic acids (formic, lactic, acetic, propionic and butyric acid) and alcohols (methanol, ethanol); in gaseous phase it is CO$_2$ and H$_2$. The drop in pH value is typical for the formation of acids due to the production of acids and metabolites with a strong odour (propionic, butyric acid, etc.)

3. Methane bacteria can use only acetic and formic acid, methanol, CO$_2$ and H$_2$ as the substrate. Higher organic acids (propionic, butyric, etc.) are transformed into acetic acid in the so-called acidogenic phase. Due to energy reasons, aceticogenic bacteria have to live in a spatially tight symbiosis with methane bacteria (or sulphate reducers). The degradation of propionic or butyric acids into acetic acid can only take place if we constantly remove the products of the reaction - H$_2$, i.e. if we maintain a low partial hydrogen pressure.

4. The methanogenic phase (production of methane) takes place by means of methane bacteria under strictly anaerobic conditions. We can divide them into two types: the first type uses H$_2$, while the second reduces CO$_2$ into methane CH$_4$. Bacteria can gain relatively high levels of energy (135kJ/mol CH$_4$), grow relatively fast and are, in comparison with the second group, pH tolerant. The second group includes less types of bacteria; they produce methane from acetic acid. The amount of energy obtained in this reaction (31kJ/mol CH$_4$) is substantially lower than with the degradation of the above-listed substrates. Methane bacteria which degrade acetic acid (acetoclastic) are very important, because they produce approximately 2/3 of methane. These bacteria grow slowly, in comparison with the previous type and the constancy of the living conditions is very important. In most cases they determine the efficiency and stability of anaerobic digestion in biogas plants (Stafford et al., 1979).

5.4 Expected anaerobic digestion of soapsuds in the biogas plant

Working on the premise that practically all organic matter is decomposed in the biogas plant, and taking into consideration the data from available literature, we conclude that soapsuds will decompose in the biogas plant:

- Sodium or sodium ions are present as a remnant of the decomposition of soap and as a remnant of sodium hydroxide NaOH which was added in the crude oil neutralisation phase. The content of sodium per 7 t of soapsuds is 81.9 kg which means 11.7 mg/1000 mg. Sodium (NaOH, Na$_2$CO$_3$, NaHCO$_3$) is important as a neutralization agent for balancing pH, because it enables us to solve the problem of precipitation which occurs if we use lime (CaO or Ca(OH)$_2$) as a neutralization agent as soon as the solubility limit is reached, i.e. the solubility constant of CaCO$_3$ is 4.96 x 10^-9 (Carbon dioxide…). The discarded calcium carbonate frequently builds up in the anaerobic sludge due to its good sedimentary properties; partially, it leaves the cleaning system as an inorganic suspended load transport at the water outlet. Lime is used because it is cheap. According to Baras (1979) Na+ in the amount of 100 - 200 mg/l has a stimulatory effect on microorganisms in anaerobic digestion; only in higher concentrations, above 350 mg/l Na+, does it act as an inhibitor.

- Phosphor or phosphorus ions are present as a remnant of the decomposition of phosphatides and as a remnant of phosphoric (V) acid H$_3$PO$_4$ which was added in the crude oil degumming phase. The total content of phosphorus is 31.9 kg per 7 t of soapsuds, which is 4.6 mg/1000 mg. To an extent, phosphorus (PO$_4$ 3-) is used as a nutrient for bacterial growth (Nekrep, 2004).

- The water temperature is around between 33 – 40°C, assuming that we choose anaerobic digestion under the mesophilic operating conditions. According to Zupančič (2005), the anaerobic digestion takes place in three temperature ranges: the psychrophilic digestion at temperatures between 5 - 25°C, the mesophilic digestion at temperatures between 33 - 40°C, and the thermophilic digestion at temperatures of 50 – 60°C. Anaerobic digestion process rate of reactions increases at higher temperatures, therefore as regards wastewater treatment technology the mesophilic digestion and the thermophilic digestion seem the most interesting; the temperatures between 5 - 25°C allows for extremely slow anaerobic digestion. Thermophilic digestion can be up to eight times faster and more efficient than mesophilic digestion. However, the reactors for anaerobic digestion which operate in the thermophile range are considered more unstable and require greater investment of energy, so they are less frequently used (Zupančič et al., 2005). Nekrep (2004) writes that methanogenic microorganisms which determine the speed, efficiency and operation of the biogas plant are adapted to a certain temperature. Even small variations in temperature cause a substantial decrease in activity. Therefore the temperature should be kept exactly within a range of +/- 1-2°C. Therefore, we can claim that the temperature really will be within the range described.

- As expected, pH values in the biogas plant is held between 6.6. and 8.0, because, according to Nekrep (2004), pH values out of 6.6. and 8.0 in an alkaline or acidic direction strongly limits the efficiency of the process. Therefore, the metabolism is affected directly and/or indirectly by changing the dissociation level of an entire species of metabolites. The range of stable methane production is
very narrow. If we want to achieve a stable anaerobic digestion process, we have to maintain this pH value.

5.5 Wastewater at the outlet

The Decree on the emission of substances and heat in the discharge of wastewater into waters and the public sewage system (Official Gazette of the Republic of Slovenia, no. 47/05 and 45/07) in Annex 1 and 2 determines limit parameters which must not be exceeded when discharging wastewater. For the above-mentioned soapsuds we can assume that the wastewater discharged from the biogas plant contains sodium ions and phosphates unless the phosphates are not partially or completely used up as a nutrient for bacterial growth. In terms of reference, the following parameters of permissible concentrations in the discharged water (in our case) were prescribed:

Table 3: Permissible concentrations in the discharged water (Verbič, 2008).

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Unit</th>
<th>Maximum permissible level of discharge into waters</th>
<th>Maximum permissible level of discharge into the public sewage system</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sodium</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Phosphorus</td>
<td>mg/l</td>
<td>2</td>
<td>--</td>
</tr>
<tr>
<td>Temperature</td>
<td>°C</td>
<td>30</td>
<td>40</td>
</tr>
<tr>
<td>pH</td>
<td></td>
<td>6.5 – 9.0</td>
<td>6.5 – 9.5</td>
</tr>
</tbody>
</table>

Moreover, the decree does not define the maximum allowed concentration of sodium.

The pH and temperature are within the prescribed limits. An exception would be a possibly higher temperature if, at the biogas plant outlet, wastewater was discharged directly into waters. In practice, the wastewater at the outlet (prior to being discharged into waters or the public sewage system) is usually used to heat the incoming substrate for the biogas plant, which consequently means the wastewater is cooled and savings are made on heating the incoming substrate.

The data presented for sodium and phosphorus ions refer only to the amounts in 7 t of soapsuds (100 t of vegetable oil processed), but do not take into consideration any possible dilution in the biogas plant when, in a bigger biogas plant (1 MW of electrical energy generated per hour), different substrates are processed at the same time (up to 250 t) each day).

It is important to emphasise that, according to the data in the available literature, wastewater from the biogas plant is, as a rule, sent in the aerobic wastewater treatment plant, because complete wastewater treatment can only be performed in connection with subsequent aerobic digestion (Baras et al., 1979).

5.6 Sludge (biomass)

Sludge (biomass), which is also produced in the biogas plant, is essential for the operation of the plant. Only excess sludge is removed. When assessing the effect of sludge removed from
the biogas plant, our premise is the data regarding the amount of substances in sludge which are produced when 7 t of soap-suds are processed (per 100 t of crude vegetable oil).

The terms of reference for processing biodegradable waste into compost (Official Gazette of the Republic of Slovenia, no. 42/04) state in Article 4 that only biodegradable waste can be composted, as described in Annex 1 to these terms of reference. These substances include waste and sludge from wastewater treatment plants, with the exception of waste and sludge of animal origin and from slaughterhouses. Article 10 states that compost is suitable for use as a plant nutrient if, with regard to the percentage of undesired substances, microbiological requirements and content of hazardous substances, it complies with the requirements of the regulation on emissions into soil.

The decree on the limit input concentration values of hazardous substances and fertilizers in soil (Official Gazette of the Republic of Slovenia, no. 84/2005) in the first item of Article 2 defines emissions into soil or onto the soil as the discharge, dumping or storage of hazardous substances or fertilizers when sludge is emitted from wastewater treatment plants, compost or silt from riverbeds and lakes and when irrigating plants or fertilizing with manure or mineral fertilizers. The fifth item of Article 2 defines as hazardous the substances or groups of substances presented in table 1 of Annex 2 and in table 8 of Annex 3 of the same decree, which are emitted into soil by means of sludge from wastewater treatment plants, compost or silt, or when irrigating plants, and which due to their properties, quantity or density, negatively affect the use of the soil or the quality of the waters. In the sixth item of Article 2 fertilizers are defined as any kind of substances which contain nitrogen compounds or potassium and which, by emission into the soil, encourage the growth of plants. Fertilizers can be of animal or mineral origin, from sludge coming out of wastewater treatment plants, compost or silt from riverbeds and lakes and when irrigating plants or fertilizing with manure or mineral fertilizers. The term of reference for processing biodegradable waste into compost (Official Gazette of the Republic of Slovenia, no. 42/04) state in Article 4 that only biodegradable waste can be composted, as described in Annex 1 to these terms of reference. These substances include waste and sludge from wastewater treatment plants, with the exception of waste and sludge of animal origin and from slaughterhouses. Article 10 states that compost is suitable for use as a plant nutrient if, with regard to the percentage of undesired substances, microbiological requirements and content of hazardous substances, it complies with the requirements of the regulation on emissions into soil.

In the Annexes of the above decree regarding the annual limit of input of dangerous substances, the maximum value of hazardous substances in sludge, silt or compost are defined. The products resulting from the decomposition of soapsuds, are not included, with exception of the following parameters:

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Unit</th>
<th>Annual maximum intake level</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sodium (in irrigation water)</td>
<td>mg/l</td>
<td>70</td>
</tr>
<tr>
<td>Phosphorus as P&lt;sub&gt;2&lt;/sub&gt;O&lt;sub&gt;5&lt;/sub&gt; (livestock manure)</td>
<td>kg/ha</td>
<td>120</td>
</tr>
</tbody>
</table>

Taking into consideration the terms of reference and regulations, all substances and all products resulting from the decomposition of soapsuds, even in the worst case scenario, will remain in the sludge. The data only refers to the amount of substances in 7 t of soapsuds (100 t of processed vegetable oil), without taking into consideration possible dilution when, in a bigger biogas plant (1 MW of electrical energy generated per hour), different substrates are processed at the same time (up to 250 t) each day. Therefore, we do not expect the sludge to present an environmental burden.

5.7 Effect of soapsuds on the operation of the biogas plant and the environment

With regard to the findings and, assuming that organic matters can decompose, we assume that soapsuds should not have a negative impact on the operation of the biogas plant, and that the products (parameters) resulting from decomposition are more acceptable for the environment than products occurring in the process used up to now.

When it comes to the anaerobic degradation of soapsuds in the biogas plant, we ought to mention several other advantages:

- Soapsuds do not contain nitrogen, sulphur or their compounds such as ammonium ions and hydrogen sulphide H2S. Although in lesser quantities sulphur and nitrogen are important for renewal of the biomass, according to several authors their compounds in certain concentrations, or products resulting from their degradation, may have a toxic effect on microorganisms in the anaerobic digestion process. Balancing their concentration so as to keep them below the toxic limit is difficult (Baras et al., 1979, 1979, Gray, 1992, Nekrep, 2004).
- Furthermore, soapsuds contain no heavy metals such as Cd, Cr, Cu, Pb, Ni, Zn, As, Fe, Mn, Hg, Ag, Al, Co, Mo, Se, Sn, which are, in low concentrations, important microelements. The majority of them are necessary for the renewal of the biomass as trace elements; however, they are toxic in higher concentrations (Deublein and Steinhauser, 2008).
- The advantage of the anaerobic digestion of soapsuds in the biogas plant lies also in the fact that soapsuds are composed of proteins and fats provides a higher amount of methane. (Eder and Schulz, 2006).

The final answer regarding the effect of soapsuds on the operation of biogas plants and on the pollution of environment can be obtained through practical testing with different incoming substrates in individual biogas plants.
5.8 The use of biogas

With the system for the anaerobic digestion of biodegradable waste we can reverse the trend of slowly accumulating waste materials and the rising cost of their removal. At the same time, we gain heat for heating and/or electrical energy which can be used on location or can be sold to customers (Burns, 2007). The literature also mentions a third option, i.e. gas processing (purification, comprimming) and sending biogas to the gas network (due to the fluctuating volume and quality of biogas this would only be possible as an exception).

6 Conclusion

The obligatory process of soapsuds remake (soapsuds as a secondary product) is performed in companies which process vegetable oils. This process is necessary due to the requirements of the ecological legislation, as soapsuds are a waste material which is not acceptable for the environment.

The currently applicable obligatory process of soapsuds remake (batch soap splitting) produces technical fatty acids which are sold to the cosmetics industry, as well as wastewater which has to be sent in the aerobic wastewater treatment plant. This process produces another waste material, calcium sulphate, which has to be transported to a special waste landfill site.

The aim of this article was to present how it is possible to create this kind of process, apply it in practice or create it in a manner which proves to be the most beneficial in terms of added value for the business system. Should this solution be applied, i.e. dispatching soapsuds, this solution would significantly help to cut waiting, preparation and finalisation times, as well as the amount of time taken by the process itself.

This means no more costs for chemicals, lower wages, no more depositing calcium sulphate in special waste landfills, using less electrical energy, less taxes for the environmental pollution and, finally, no more technical equipment for separating soapsuds. The latter is especially important for the new technological vegetable oil processing lines or in plants with old and worn out equipment.

This would result in changes to the organizational structure, the human resources system, systematization of work positions and the information system. It would be sensible to implement computer-managed measurement of the quantities of soapsuds in the reservoirs, which means the line manager of the refinery would be able to monitor the level of supply on a computer screen at any time. This would additionally simplify the process.

It is important to emphasize that innovation and process reengineering contribute towards sustainable development, since environmental care, due to global environmental changes and the ever more stringent ecological legislation, has become one of the ongoing tasks and priorities of the management.

The advantage of anaerobic digestion of soapsuds in the biogas plant lies also in the fact that soapsuds are composed of soap (water solution), which makes the hydrolysis phase unnecessary in the biogas plant, except when neutral oil is present. This shortens the degradation time. With the system for the anaerobic digestion of biodegradable waste we can reverse the trend of slowly accumulating waste materials and increasing costs for their removal. At the same time, we gain heat for heating and/or electrical energy which can be used on location or can be sold to customers.

Based on these findings, we can conclude that it is feasible to decomposed soapsuds in the biogas plant and that the decomposition of soapsuds has no negative impact on the operation of the biogas plant. In compliance with the requirements of the applicable ecological legislation, products (parameters) resulting from this decomposition present less pollution of the environment than the products occurring in the process currently in use.

Literature:


Prenova in inovacija procesa predelave milnice


Pri reševanju tega okoljskega problema je prišlo do ideje, da milnice ne bi več predelovali, pač pa jo odvažati na bioplinsko napravo, kjer se razkroji do bioplina. Bioplin se lahko uporabi za pridobivanje električne energije, ogrevanje ali dodajanje k mestnemu plinu. Prenova procesa je v tem primeru tudi inovacija procesa in inovacija s področja ekologije.

Ključne besede - prenova poslovnega procesa, inovacija procesa, predelava milnice, bioplinska naprava, ekologija